

Introduction to General, Organic and Biochemistry II (Chem. 30B.01) Syllabus

Lecture: Tu-Th 5:30 PM – 7:20 PM -- Room G1

Lab: Wed 7:30-10:20 PM -- Room SC2210

Instructor: Dr. James Maxwell, Mobile phone: (773) 454-7779 (texts also), email: maxwelljames@fhda.edu

Office Hours: Tuesday & Thursday: 4-5pm, Second Floor SC1

Description: This class is for students entering the allied health fields. The focus of the second part of Introduction to General, Organic, and Biochemistry is organic and biochemistry. The topics included in organic chemistry are: hydrocarbons, alcohols, thiols, ethers, carboxylic acids, esters, amines, and amides. Various physical and chemical properties of these organic substances will be studied along with nomenclature and structural features. The topics included in biochemistry are: carbohydrates, fatty acids and lipids, amino acids and proteins, nucleic acids and DNA. Various physical and chemical properties of these biological molecules will be studied. A brief introduction to metabolism will also be discussed.

Prerequisites: Chemistry 30A or 25 or 1A. Advisory: English Writing 211 and Reading 211 (or Language Arts 211), or English as a Second Language 272 and 273.

Student Learning Outcomes:

1. Solve stoichiometric problems by applying appropriate molar relationships.
2. Predict the behavior of ideal gasses using Kinetic Molecular Theory.
3. Apply acid-base chemical principles to biological processes.

Evaluation: Your grade will be based on your performance in the following:

8 best Quizzes (10 pts each)	80 points
9 Labs (20 pts each)	180
Cleanup Crew	50
Genetic Disorder Paper	100
3 Exams (100 pts each)	300
<u>1 Final (200 pts)</u>	<u>200</u>
Total	910 points

Letter grades will be assigned according to the *approximate* scale:

A	90%
B	80%
C	70%
D	50%
F	< 50%

Attendance: Your attendance is urged for all lectures and required for all quizzes, exams and labs. Unexcused exam, quiz and lab absences score 0. It is the responsibility of the student to contact the professor regarding missed work. If an absence is anticipated, the student should make arrangements to complete the missed assignments prior to the absence. In an emergency, it is the student's responsibility to contact the instructor within one class period of an exam. *There are no laboratory make-up days.* Please sign the attendance sheet each class.

Quizzes: Quizzes will be given during class on Monday or Wednesday as scheduled in syllabus, and will have a time limit. Answer keys will be available after the quiz. *If you miss the quiz, you will **not** have a chance to make it up.* The best 10 quiz scores will be used in determining your final grade.

Exams: There will be three exams and one final exam. You must bring your own calculator (if you need one), pencil and eraser for exams. You are permitted to bring a molecular model kit, the instructor must approve if it is assembled

in any way. Cell phones may not be used at any time during the exam. Calculators may be used if approved by instructor. Once the exam begins you may not leave the room unless you turn in the exam, so plan to take a bathroom break *before* class. **No Cell Phones during Exam! Answer Keys will be available after the exam.**

Text: **Chemistry** by Karen C Timberlake, 12e, 2015, Pearson. You may use another edition if you have it, but you are responsible for know the differences and what material may be omitted for your copy of the text. For textbook bargain prices check out textbooksrus.com, half.com or Amazon marketplace for used books. You might be able to save a bunch of \$\$\$. You can rent your text at www.vitalsource.com.

Labs: All 9 labs count towards your grade. No make-up labs. Late labs will incur a penalty. You **MUST** wear eye protection during lab!

Academic Dishonesty: "Academic dishonesty is a serious offense, which includes but is not limited to the following: cheating, complicity, fabrication and falsification, forgery, and plagiarism. Cheating involves copying another student's paper, exam, quiz or use of technology devices to exchange information during class time and/or testing. It also involves the unauthorized use of notes, calculators, and other devices or study aids. In addition, it also includes the unauthorized collaboration on academic work of any sort. Complicity, on the other hand, involves the attempt to assist another student to commit an act of academic dishonesty. Fabrication and falsification, respectively, involve the invention or alteration of any information (data, results, sources, identity, and so forth) in academic work. Another example of academic dishonesty is forgery, which involves the duplication of a signature in order to represent it as authentic. Lastly, plagiarism involves the failure to acknowledge sources (of ideas, facts, charges, illustrations and so forth) properly in academic work, thus falsely representing another's ideas as one's own."

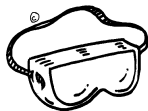
Word Processing: If you are looking for a **free** word processor compatible with WORD, checkout www.openoffice.org .

Online Help: Some suggested websites for help. <http://www.general-chemistry-help.com>
http://chemistry.about.com/od/chemistry101/Chemistry_101_Introduction_to_Chemistry.htm
Mastering Chemistry (sold with and in addition to) for Timberlake Text. I'm not familiar with this material. We can discuss it in class to see if it is worth the cost.

Help: If you need help with any aspect of this course, please contact your instructor first. You can also contact the Student Success Center at <http://www.deanza.edu/studentssuccess/> to get help with tutoring or with reading, and writing, tutoring or academic skills. Please use this resource.

Calculator: You still need a simple scientific calculator not associated with your mobile phone. They will cost about \$10.00.

Eye Protection: You must wear full goggles and not safety glasses. Without them, you may not participate in lab and will receive a grade of zero for that lab. See illustration below. They are available at the bookstore or at hardware stores.



Changes to Syllabus: **This syllabus may change according to the instructor and the needs of the class.**
Please check with the syllabus posted

Date Tues	Lecture	Date Wed	Lab	Date Thr	Lecture
5 Apr	Intro to Course and Lab Lecture: Ch. 11: Intro to Org. Chem.: Hydrocarbons	6 Apr	Check-In	7 Apr	Lecture: Chap. 11: Cont.
12 Apr	Lecture: Chap. 12: Alcohols, Thiols, Ethers, Aldehydes, and Ketones Quiz 1: Chap 11	13 Apr	Lab 1: Hydrocarbons Signed Safety Document due	14 Apr	Lecture: Chap. 12: Cont.
19 Apr	Lecture: Chap. 13: Carbohydrates Quiz 2: Chap 12 Topic Choice for Paper Due!	20 Apr	Lab 2: Alcohols and Phenols	21 Apr	Lecture: Chap. 13: Cont.
26 Apr	Lecture: Review Chap. 11-13 Quiz 3: Chap 13	27 Apr	Lab 3: Aldehydes and Ketones	28 Apr	EXAM 1: Chap 11-13
3 May	Lecture: Chap. 14: Carboxylic Acids, Esters, Amines, and Amides	4 May	Lab 4: Carboxylic Acids and Esters	5 May	Lecture: Chap. 14: Cont.
10 May	Lecture: Chap. 15: Lipids Quiz 4: Chap 14	11 May	Lab 7: Amines and Amides	12 May	Lecture: Chap. 15: Cont.
17 May	Lecture: Chap. 16: Amino Acids, Proteins, and Enzymes Quiz 5: Chap 15	18 May	Lab 5: Carbohydrates	19 May	Lecture: Chap. 16: Cont.
24 May	Lecture: Review Chap. 14-16 Quiz 6: Chap 16	25 May	Lab 6: Glycerphospholipids and Steroids	26 May	EXAM 2: Chap. 14-16
31 May	Lecture: Chap. 17: Nucleic Acids and Protein Synthesis	1Jun	Lab 8: Acetaminophen	2 Jun	Lecture: Chap. 17: Cont.
7 Jun	Lecture: Chap. 18: Metabolic Pathways and Energy Production Quiz 7: Chap 17 Paper DUE!	8 Jun	Lab 9: Peptides and Proteins	9 Jun	Lecture: Chap 18: Cont.
14 Jun	Lecture: Final Review Chap 11-18 Quiz 8: Chap 18 Quiz 9: Mini Final Chap 11-14 (Take-Home) Quiz 10: Mini Final Chap 15-18 (Take-Home)	15 Jun	Lab: Final Exam Check-Out	16 Jun	Exam 3: Chap 17 & 18 Quiz 9 & 10 DUE
21 Jun	Final Exam: Chap 10-18 @ 6:15 pm	22 Jun	No Class	23 Jun	No Class

INSTRUCTIONS for the Laboratory:

1. Print out, read, sign and return to your instructor the safety statement in the link below. This must be returned by the second laboratory period 13 April, 2016).
[https://myportal.fhda.edu/cp/grouptools/filesshare/24512/93057/Student%20Safety%20Document\(2\).pdf](https://myportal.fhda.edu/cp/grouptools/filesshare/24512/93057/Student%20Safety%20Document(2).pdf)
2. You must do your laboratory work at the time assigned. Attendance will be taken. Study the experiment carefully before coming to class so that you don't waste time going through the procedure during the lab. **NO MAKE UP LABS.**
3. You must do your own work unless you are told to work in pairs for an experiment. If you need guidance, let the instructor know.
4. Always think through the next step you are going to perform before starting it.
5. **Record all data in ink while you work.** Do not write data on paper towels or other pieces of paper, even temporarily. Make sure your data is complete. **Do not forget to write your name or record any unknown number.** Pay attention to significant figures and units. If you make a mistake, cross it out neatly with a **single** line.
6. All laboratory reports are due one week after the experiment is performed.
7. Children are not allowed in the lab.
8. No eating or drinking in the lab.
9. **ALWAYS WEAR YOUR EYE PROTECTION.** Failure to wear your eye protection will lead to dismissal from lab and a lowered grade for that experiment.
10. You **MUST WEAR LONG PANTS** and **SENSIBLE CLOTHING** when we are doing any lab that required Safety Goggles as discussed during the safety lectures. This is a school policy. If you wear shorts, sandals, or other clothing that is not consistent with safety, you will **not** be admitted to the laboratory. Wear a lab apron if you have one. **You can NEVER WEAR SHORT PANTS or SKIRTS during LABORATORY PERIODS.**
11. Always work with clean equipment. Clean also means **DRY**.
12. Carefully measure the quantity of each material to be used in the experiment.
13. Always place reaction vials, test tubes or flasks in a clean beaker when standing them on a laboratory bench.
14. Do not take reagent bottles to your laboratory work area. Use test tubes, beakers, or paper to obtain chemicals from the dispensing area. Take small quantities of reagents. You can always get more if you run short.
15. Check carefully the label on each reagent bottle to be sure you have the correct reagent. The names of many substances appear similar at first glance.
16. To avoid possible contamination, never return unused chemicals to the reagent bottles. Never interchange spatulas or droppers.
17. Do not insert droppers into large reagent bottles. Instead pour a little of liquid into a small beaker.
18. Be neat in your work; if you spill something, clean it up immediately.
19. Wash your hands anytime you get chemicals on them and at the end of the laboratory period.
20. Keep the mass balances and the area around them clean. Follow the directions given by the instructor on the proper weighing technique to use. Otherwise, do not place chemicals directly on the balance pans; place a piece of weighing paper or a small container on the pan first, and then weigh your material. Never weigh an object while it is hot.
21. Do not heat graduate cylinders, burettes, pipettes, or bottles with a burner flame.
22. Do not look down into the open end of a test tube in which the contents are being heated or in which a reaction is being conducted.
23. Do not perform unauthorized experiments.
24. After completing the experiment, clean and put away your glassware and equipment. Clean your work area and make sure the gas and water are turned off. A clean lab is a safe lab.
25. Dispose solid waste such as filter paper, litmus paper, and matches in the wastebasket, not in the sink. Dispose broken glass in the broken glass waste boxes. Dispose all other solid chemicals as directed by your instructor. Pour all the toxic liquids into the waste bottles provided or as directed by instructor.
26. **Cleanup Crew:** After each lab a crew of two or three students will be assigned cleanup duties for a total of 50 for points each member of the crew. Failure to perform all cleanup procedures will result in loss of points equally for all crew members. If a crew member is absent, they will be reassigned another lab.
27. **Cleanup Crew Duties:**
 - a. **Balance room clean of all debris and all doors on all balances closed.**
 - b. **Balanced room closed and locked,**
 - c. **All reagents used during lab are wiped down, lids secured and replaced in proper area as found at beginning of lab.**
 - d. **All table tops cleaned and dry; floor is free of paper and spills.**
 - e. **All stray paper and water spills cleaned.**
 - f. **All Distilled water bottles refilled.**
 - g. **Waste containers are properly labeled (review with instructor), wiped clean, lids securely closed and placed in proper secondary containment.**
 - h. **Projector is turned off, screen is up and white board is erased.**

Genetic Disorder Paper

1. Topic must be chosen and approved from list given on Course Studio web page
2. Paper must be 3 full pages, double spaced, 1 inch margins.
3. There must be at least three references, one must be a book, magazine article or newspaper article. References will be listed on a separate page.
4. Paper must be written in proper English and spelling. Such mistakes will be counted against the total points.
5. The Paper must include the following:
 - a. Name of Disorder
 - b. Nature of Disorder: body systems effected
 - c. Discoverer of Disorder, Date of Discovery
 - d. Brief Bio of Discoverer
 - e. Symptoms
 - f. Treatment/Cure
 - g. Prognosis of patients
 - h. Future of disease
6. Paper must be printed on plain white paper, Cover sheet is recommended.
7. Any illustrations must not be counted as part of the body of the paper and shown on pages separate from the text of the paper.
8. For the best grade, the paper must be interesting and compelling. The best paper in the class will receive the best grade.